

HOW MUCH CHARGE IN
A MOLE OF ELECTRONS?

→ LIKE A "DOZEN"
→ 6.02×10^{23}

$$q_e = -1.6 \times 10^{-19} \text{ C}$$

$$Q = q_e \times N_0 = 96320 \text{ C}$$

Jan 26-12:00 PM

CARBON

- LOOK UP ATOMIC #

Jan 26-12:59 PM